

On the shape of giant soap bubbles

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We study the effect of gravity on giant soap bubbles and show that it becomes dominant above the critical size $\ell = a^2/e_0$, where e_0 is the mean thickness of the soap film and $a = \sqrt{\gamma_b/\rho g}$ is the capillary length (γ_b stands for vapor–liquid surface tension, and ρ stands for the liquid density). We first show experimentally that large soap bubbles do not retain a spherical shape but flatten when increasing their size. A theoretical model is then developed to account for this effect, predicting the shape based on mechanical equilibrium. In stark contrast to liquid drops, we show that there is no mechanical limit of the height of giant bubble shapes. In practice, the physicochemical constraints imposed by surfactant molecules limit the access to this large asymptotic domain. However, by an exact analogy, it is shown how the giant bubble shapes can be realized by large inflatable structures.

soap bubbles | Marangoni stress | self-similarity

Soap films and soap bubbles have had a long scientific history since Robert Hooke (1) first called the attention of the Royal Society and of Newton to optical phenomena (2). They have been of assistance in the development of capillarity (3) and of minimal surface problems (4). Bubbles have also served as efficient sensors for detecting the magnetism of gases (5), as elegant 2D water channels (6), and as analog “computers” in solving torsion problems in elasticity (7, 8), compressible problems in gas dynamics (9), and even heat conduction problems (10). Finally, in the last decades, the role of soap films and bubbles in the development of surface science has been crucial (11–13), and the ongoing activity in foams (14, 15) and in the influence of menisci on the shapes of bubbles (16) are modern illustrations of their key role. The shape of a soap bubble is classically obtained by minimizing the surface energy for a given volume, hence resulting to a spherical shape. However, the weight of the liquid contained in the soap film is always neglected, and it is the purpose of this article to discuss this effect.

For liquid drops, the transition from a spherical cap drop to a puddle occurs when the gravitational energy, $\rho g R^4$ (R is the typical size of the drop), becomes of the same order as its surface energy $\gamma_b R^2$. That is, for a drop size of the order of the capillary length $a = \sqrt{\gamma_b/\rho g}$ (γ_b is the liquid–vapor surface tension, and ρ is the liquid density). Typically, this transition is observed at the millimetric scale: For a soap solution with $\gamma_b = 30$ mN/m, $a \approx 1.7$ mm. The two asymptotic regimes may be distinguished through the behavior of the drop height h_0 with volume: $h_0 \approx R$ for small spherical drops, while the height of large puddles saturates to a constant value $h_0 \approx a$.

If we look for the same transition in soap bubbles, we expect the gravitational energy, $\rho g R^3 e_0$, to become of the order of the surface energy, $\gamma_b R^2$, at the typical size $R \approx \ell$, with $\ell = a^2/e_0$ (e_0 stands for the mean thickness of the film). Thanks to the iridescence, the mean thickness can be estimated to a few microns or less, and the light–heavy transition is thus expected at the metric scale (instead of the millimetric one for drops): For $\gamma_b = 30$ mN/m and $e_0 = 1 \mu\text{m}$, $\ell \approx 3.1$ m.

The experimental setup dedicated to the study of such large bubbles is presented in *Experimental Setup*, before information on *Experimental Results and Model*. The discussion on the asymptotic shape and the analogy with inflated structures is presented in *Analogy with Inflatable Structures*.

Experimental Setup

The soap solution is prepared by mixing two volumes of Dreft[©] dishwashing liquid, two volumes of water, and one volume of glycerol and was left aside for 10 h before experiments. The surface tension of the different mixtures was measured using the pendant drop method. It was found to be $\gamma_b = 26 \pm 1$ mN/m.

The bubbles are formed in a round inflated swimming pool of 4 m diameter (Fig. 1), filled with 10 cm to 20 cm of soap solution. A large bubble wand was assembled with two wood sticks and two cotton strings. The strings were immersed in the soap solution. Two experimenters, located on opposite sides of the pool, slowly opened the loop in air and pulled the sticks above the water surface, before dipping the loop into the water to form the bubble.

Once the bubble is at rest, the shape is analyzed by side view images as shown in Fig. 1. In particular, we measure the diameter, $2R$, and the height, h_0 , of the giant bubble. The camera is placed 4 m from the center of the pool, and the center of the lens is at the same height as the center of the bubble, to minimize parallax errors.

The film thickness e_0 is important when studying the effect of the bubble weight, as it determines the liquid mass. Following McEntee and Mysels (17), the thickness of the bubble is measured via the bursting technique: A hole in a punctured soap film

Significance

Surface tension dictates the spherical cap shape of small sessile drops, whereas gravity flattens larger drops into millimeter-thick flat puddles. In contrast with drops, soap bubbles remain spherical at much larger sizes. However, we demonstrate experimentally and theoretically that meter-sized bubbles also flatten under their weight, and we compute their shapes. We find that mechanics does not impose a maximum height for large soap bubbles, but, in practice, the physicochemical properties of surfactants limit the access to this self-similar regime where the height grows as the radius to the power 2/3. An exact analogy shows that the shape of giant soap bubbles is nevertheless realized by large inflatable structures.

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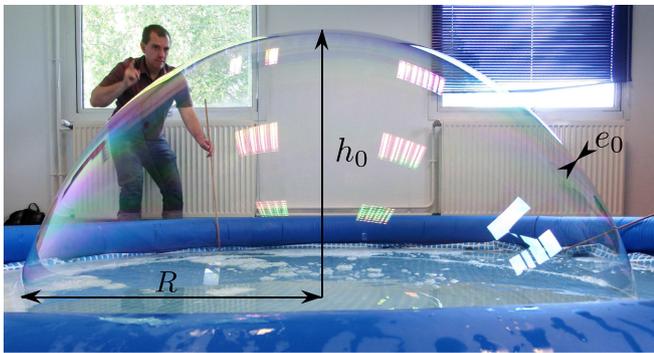


Fig. 1. Presentation of a “giant” soap bubble and definition of its radius, R , and height, h_0 . (Here, $R = 1.09$ m, and $h_0 = 0.97$ m.)

grows because of unbalanced surface tension forces at the edge of the hole. The opening velocity v is constant and is given by the Dupré–Taylor–Culick law (18–20),

$$v^2 = \frac{2\gamma_b}{\rho e_0} = 2g\ell, \quad [1]$$

where $\gamma_b \approx 26$ mN/m, and $\rho \approx 1,000$ kg·m⁻³. Examples of thickness measurements are presented in Fig. 2: In Fig. 2A and Fig. 2B, we present two sequences of four pictures showing the opening of a hole in two soap bubbles of different sizes. We use such sequences to extract the bursting velocity plotted as a function of time in Fig. 2C. We observe that v is almost constant and takes the value of 8 m/s for sequence in Fig. 2A and 2.8 m/s for sequence in Fig. 2B. From the value of v , we deduce $e_0 \approx 0.81$ μm and $\ell \approx 3.2$ m for Fig. 2A and $e_0 \approx 6.6$ μm and $\ell \approx 0.4$ m for Fig. 2B, using Eq. 1. Although it may seem surprising to find that the film thickness is homogeneous, earlier studies on the drainage of almost spherical liquid shells have shown that the thickness approaches a profile with little spatial variations (21, 22).

Experimental Results

Once ℓ is determined, we use it to rescale the experimental shapes. An example of a nonspherical bubble is presented in reduced scale in Fig. 3A. A systematic analysis of the effect of gravity on the bubble shape is shown in Fig. 3B, where we plot the reduced height h_0/ℓ as a function of the bubble reduced radius R/ℓ for all experiments. Fig. 3B reveals that the bub-

bles’ shapes remain approximately spherical ($h_0/\ell = R/\ell$, black dashed line) only up to $R/\ell \approx 0.3$. For larger sizes, the bubble height is significantly lower than that of a sphere. The largest value of h_0/ℓ reached experimentally is ~ 1.2 , with corresponding radius $R/\ell = 1.6$. Two points must be underlined at this stage: (i) The experimental data in Fig. 3B show no sign of a height saturation for increased bubble volumes, and (ii) despite our efforts, we never managed to make bubbles larger than $R = 1$ m. Both observations will be explained in *Model*.

Model

We now consider the mechanical equilibrium of the soap film and predict how gravity affects the bubble shape. Bubbles are axisymmetric, and we assume a uniform film thickness e_0 . The bubble shape is described using the parametrization shown in Fig. 4A and B: The local height of the soap film is $h(s)$, and the local angle of the membrane relative to the horizontal is $\theta(s)$ (defined as positive everywhere). The height and angle are functions of the curvilinear coordinate s that measures the arclength starting from the top of the bubble; φ is the azimuthal angle around the vertical axis.

The equilibrium of an infinitesimal part of the membrane of surface area $rdsd\varphi$ is first considered along the s direction. To account for the experimentally observed slow draining and long bubble lifetimes, the air–liquid interface must strongly moderate the flow and behave as partially rigid, in contrast with the no-stress behavior of surfactant-free interfaces. The main effect is that gradients in surface tension $\gamma(s)$, due to the presence of surfactants, have to balance viscous stresses applied by the flowing liquid along the interface. In reaction, viscous stresses balance the weight of the liquid in the film. Finally, the weight of liquid is fully transmitted to the walls through viscous stresses and balanced by surface tension gradients (15). The contribution due to surface tension on each side of the infinitesimal element gives a force $2\gamma r d\varphi$, where γ is a function of position s . The weight of the liquid inside the film is $\rho g e_0 r d\varphi ds \sin \theta$, when projected along the s direction. The balance of surface tension and weight thus gives

$$2r d\varphi [\gamma(s) - \gamma(s + ds)] = \rho g e_0 r d\varphi ds \sin \theta, \quad [2]$$

which, using $dh/ds = -\sin \theta$, yields

$$d\gamma = \frac{1}{2} \rho g e_0 dh \Rightarrow \gamma(s) = \gamma_b + \frac{1}{2} \rho g e_0 h(s). \quad [3]$$

Here γ_b is the surface tension of the soap solution at the base of the bubble ($h = 0$), and the surface tension is found to increase

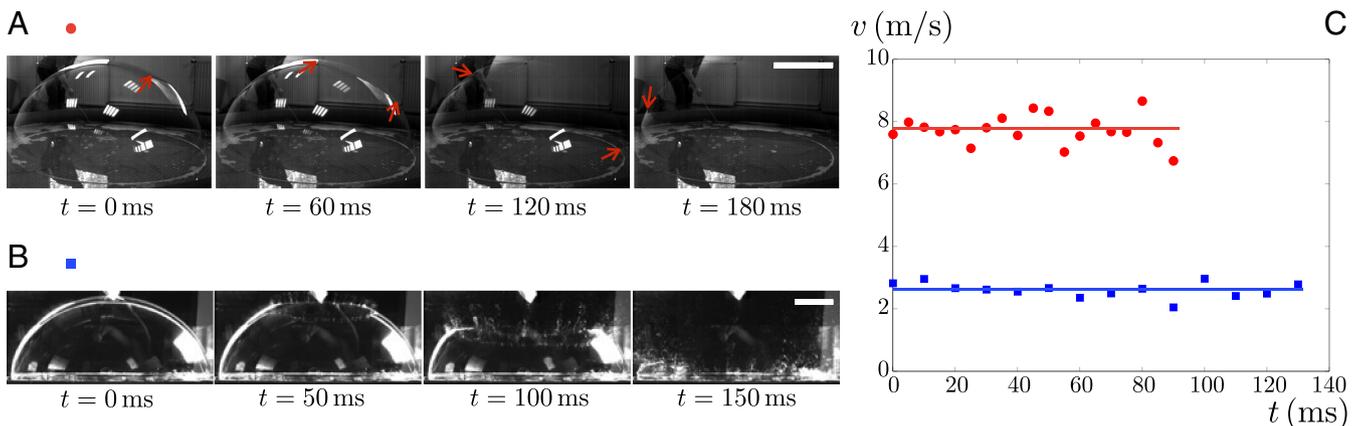


Fig. 2. Bursting of bubbles used to determine the soap film thickness. (A) Image sequence of bursting bubbles, with a time step of 60 ms between images. Red arrows indicate the boundary of the opening hole on each image. (Scale bar, 50 cm.) (B) Bursting sequence with a time step of 50 ms between images. (Scale bar, 10 cm.) (C) The bursting velocity corresponding to A and B is plotted versus time.

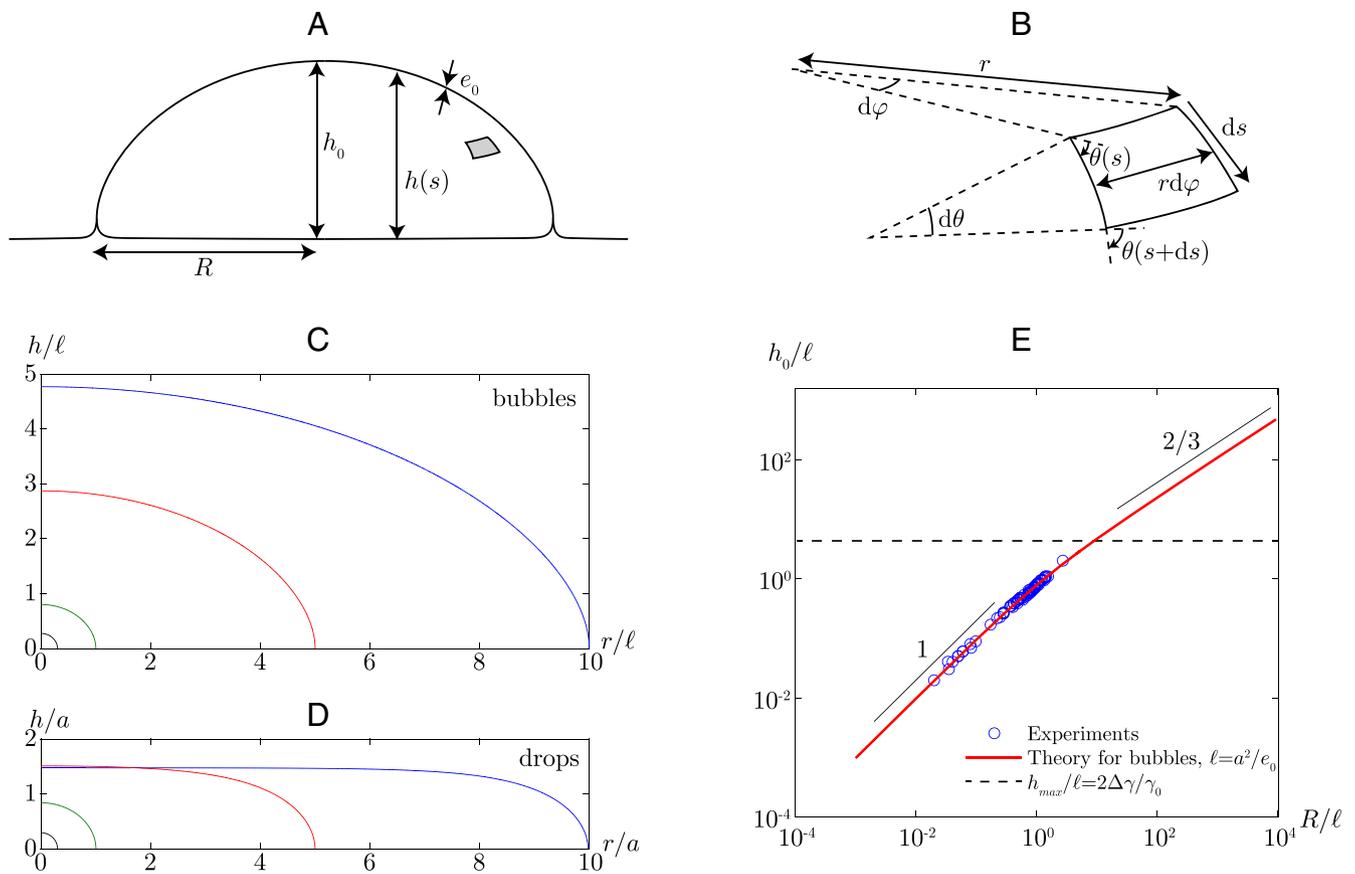


Fig. 4. (A) Sketch of a giant bubble of radius R and height h_0 . The local height is h . The gray area shows an infinitesimal surface element of the bubble. (B) Zoom on the infinitesimal part of the bubble located at distance r from the symmetry axis of the bubble. The surface area of this surface element is $r ds d\varphi$; $\tan \theta$ is the local slope of the soap film with respect to the horizontal. (C) Shapes of soap bubbles with dimensionless radius $R/\ell = 0.3, 1, 5,$ and 10 . Although large bubbles tend to flatten (i.e., $h_0/R \rightarrow 0$), the height of giant bubbles shows no sign of saturation. (D) Shapes of liquid drops with contact angle $\theta = 90^\circ$ and dimensionless radius $R/a = 0.3, 1, 5,$ and 10 . The dimensionless height of large drops saturate to $\sqrt{2}$. (E) Experimental dimensionless height of soap bubbles as a function of their dimensionless radius (circles). The theoretical dimensionless height h_0/ℓ of bubbles (full line) is plotted as a function of their dimensionless radius R/ℓ . Small bubbles ($R/\ell < 0.3$) are insensitive to gravity and remain hemispherical, thus minimizing their surface area for the given volume. Giant bubbles flatten, but there is no saturation height as exists for drops larger than the capillary length: For large bubbles, one finds $h_0 \approx R^2/3$. The physical chemistry of surfactants, however, limits the range of accessible surface tension, setting the actual upper limit for the size of giant bubbles (dashed line): $h_{max}/\ell = 2\Delta\gamma/\gamma_b$.

Fig. 4D shows the corresponding numerical solutions: Large drops develop toward puddles, which, for $\theta = \pi/2$, saturate to the height $\hat{h}_0 = \sqrt{2}$.

The height of soap bubbles may, however, be limited by physical chemistry of surfactants. The water that constitutes the bubble is prevented from draining quickly by gradients in the surface tension. The larger surface tension at the top of the bubble supports the weight of water in the liquid shell (15). In practice, the surface tension of a soap solution cannot be higher than that of pure water, γ^* ; γ also has a minimum, γ_b , set by the surfactant concentration of the solution used in the experiments. Eq. 3 thus gives a criterion for the maximal height h_{max} of the bubble,

$$\gamma^* = \gamma_b + \frac{1}{2} \rho g e_0 h_{max} \tag{8}$$

so that

$$\frac{h_{max}}{\ell} = \frac{2\Delta\gamma}{\gamma_b}, \tag{9}$$

where $\Delta\gamma = \gamma^* - \gamma_b$ is the highest achievable surface tension contrast between the top and bottom of a bubble; $\gamma^* \approx 70$ mN/m, and γ_b may typically be as low as 20 mN/m, so that the expected

maximal height of a bubble of thickness $e_0 = 5 \mu\text{m}$ is of order 2 m, close to the size of the biggest bubbles we experimentally produced (Fig. 1).



Fig. 5. Festo's Airquarium, 31 m in diameter.

Analogy with Inflatable Structures

Interestingly, the shapes we have just discussed correspond to a minimization problem that is relevant in the context of large inflatable structures, such as shown in Fig. 5. These structures consist of a thin sheet that we assume cannot be stretched, and which is inflated by a pressure difference ΔP . The mechanical analysis on an infinitesimal element of the thin sheet is strictly equivalent to that in Fig. 4 *A* and *B*: The role of surface tension is replaced by the tension that develops inside the membrane. It is interesting to confirm this analogy based on energy minimization, with the no-stretch condition imposed through a Lagrange multiplier λ . Characterizing the axisymmetric shape as $h(r)$, and thus $h' = dh/dr$, the functional $\mathcal{F}[h]$ to be minimized reads

$$\mathcal{F}[h] = \int dr 2\pi r \mathcal{L}(h, h'), \quad [10]$$

with

$$\mathcal{L}(h, h') = \rho g e_0 h (1 + h'^2)^{1/2} + \lambda (1 + h'^2)^{1/2} - \Delta P h. \quad [11]$$

The three terms respectively represent the gravitational free energy, the area constraint, and the work done by the pressure difference. The Euler–Lagrange equation for this functional gives (see *Inflatable Structures*):

$$\Delta P = (\lambda + \rho g e_0 h) \left(\frac{d\theta}{ds} + \frac{\sin \theta}{r} \right) + \rho g e_0 \cos \theta, \quad [12]$$

which is, indeed, strictly identical to the equation that dictates the bubble shapes (Eq. 4). Designing the inflatable structures along these optimal shapes will naturally avoid stretching and compression of various parts of the sheets, avoiding wrinkles and reducing tensile stresses exerted in the sheets and on the seams that connect the various parts. This design should help increase the lifetime of such structures.

Conclusion

We study the shape of large soap bubbles and show that gravity becomes important at the scale $\ell = a^2/e_0$. We derive the equation for the shape and show that gravity matters in two distinct terms: the expected hydrostatic term and the evolution of surface tension via Marangoni stresses. A direct consequence is that there is a physicochemical limit to the size of soap bubbles, h_{max} . Finally, we point out that, contrary to drops, the shape of giant soap bubbles is not characterized by a saturation of the height but by a self-similar behavior in which $h_0/\ell \approx (R/\ell)^{2/3}$.

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